



## SUBMERSIBLE MICROBIAL FUEL CELL-BASED BIOSENSOR FOR IN SITU BOD MONITORING

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### KEYWORDS

SMFC, BOD biosensor, wastewater quality.

### ABSTRACT

A biological oxygen demand (BOD) biosensor based on the microbial fuel cell (MFC) technology was tested for online and in situ monitoring of biodegradable organic content of domestic wastewater. A stable current density of  $282 \pm 23$  mA/m<sup>2</sup> was obtained with domestic wastewater containing a BOD<sub>5</sub> of  $317 \pm 15$  mg O<sub>2</sub>/L at  $22 \pm 2$  °C,  $1.53 \pm 0.04$  mS/cm and pH  $6.9 \pm 0.1$ . The current density showed a linear relationship with BOD<sub>5</sub> concentration ranging from  $17 \pm 0.5$  mg O<sub>2</sub>/L to  $78 \pm 7.6$  mg O<sub>2</sub>/L. The current generation from the SMFC biosensor was dependent on the measurement conditions such as temperature, conductivity, and pH. These results provide useful information for the development of a biosensor for real-time in situ monitoring of wastewater quality.

### INTRODUCTION

The BOD<sub>5</sub> may be the most used test to assess the amount of pollutant organic matter in water, however, is time and labour consuming, and is done ex-situ. A recent online biosensor to quantify BOD was developed using the MFC concept (Kim et al. 2003). The usual configuration was a mediator-less MFC, with two chambers separated by a cation exchange membrane and continuous wastewater flow (Gil et al. 2003, Kim et al. 2003, Kang et al. 2003, Chang et al. 2004, 2005, Moon et al. 2004, 2005). This configuration is complex and the setup is not suitable for in situ applications. Recently, a very compact MFC configuration, known as submersible microbial fuel cell (SMFC), was developed by Min and Angelidaki (2008). The aim of the present study was adapt and test the SMFC configuration as an in situ BOD<sub>5</sub> biosensor.

### MATERIAL AND METHODS

#### Biofilm formation

A biosensor was operated at room temperature ( $22 \pm 2$  °C) during 3 weeks to biofilm formation in the anodic surface. The biosensor was immersed into 1 L of

domestic wastewater contained in a glass vessel tightly and anaerobically closed. The cathode chamber was continuously flushed with air (5 mL/min).

#### Effect of BOD<sub>5</sub> concentration and environmental parameters on current density

Five wastewater dilutions were prepared with tap water ( $17 \pm 0.5$  mg O<sub>2</sub>/L -  $183 \pm 4.6$  mg O<sub>2</sub>/L) and tested until the maximum current density peak. The maximum voltage (U) was measured 2 times consecutively for each concentration across a fixed resistance (R) equal to 1 kΩ. Temperature, pH and conductivity ( $\sigma$ ) were tested independently using a diluted domestic wastewater ( $143.5 \pm 8.7$  mg O<sub>2</sub>/L,  $1.1 \pm 0.012$  mS/cm and pH  $6.5 \pm 0.2$ ). T tested varied from  $11 \pm 0.2$  °C to  $33 \pm 0.3$  °C. The wastewater pH was adjusted in the range of  $6.0 \pm 0.1$  to  $8.5 \pm 0.1$ . The wastewater  $\sigma$  varied between  $1.1 \pm 0.012$  mS/cm and  $13.4 \pm 0.013$  mS/cm.

#### Analytical methods and calculations

The organic content of wastewater was assessed both as chemical oxygen demand (COD), and BOD<sub>5</sub>, as described in Standard Methods. A correlation between BOD<sub>5</sub> and COD was experimentally determined. Then, the polarization curve (U versus current density (j)) and the power curve (power density (P) versus j) were recorded using a series of R in the range of 22 kΩ to 100 Ω. The current intensity (i) and density (j) were calculated according to the Ohm's law.

### RESULTS AND DISCUSSION

#### Biofilm formation

After a period of 3 weeks, the SMFC operated with domestic wastewater ( $317 \pm 15$  mg O<sub>2</sub>/L of BOD<sub>5</sub>) was stabilized at a maximum value of 0.27 mA ( $282 \pm 23$  mA/m<sup>2</sup>).

#### Performance of SMFC-type BOD<sub>5</sub> biosensor

The open circuit voltage measured in the stable phase of current generation (wastewater =  $298.2 \pm 4.2$  mg O<sub>2</sub>/L of BOD<sub>5</sub>) was  $414 \pm 6$  mV. The shape of the polarization curve (data not shown) confirmed the prevalence of ohmic losses generated by the resistance of membrane, electrolyte of the wastewater and,



bacterial metabolism. The maximum P obtained in the power curve (data not shown) was  $72 \text{ mW/m}^2$ , which is equivalent to a  $j = 283 \text{ mA/m}^2$  ( $1 \text{ k}\Omega$ ). These values are in agreement with the obtained in stable phase of the SMFC operation ( $1 \text{ k}\Omega$ ).

#### Current generation at various BOD<sub>5</sub> concentrations

With the different diluted wastewaters (Figure 1), the current density increased linearly with BOD<sub>5</sub> up to a concentration of  $78 \pm 8 \text{ mg O}_2/\text{L}$  ( $j = 2.68 i - 18.78$  with  $r^2 = 0.996$ ). The range of BOD<sub>5</sub> concentration that we could measure by the SMFC are in agreement by those published in literature (Gil et al. 2003, Kim et al. 2003, Chang et al 2004, Moon et al. 2004, Kim et al. 2006). The response time for a BOD<sub>5</sub> concentration of  $17 \pm 1 \text{ mg O}_2/\text{L}$  was shorter than 30 min, however, about 10 h were needed for concentrations higher than  $78 \pm 8 \text{ mg O}_2/\text{L}$ . The response time in some studies was quite low (Chang et al. 2004, Moon et al. 2004, Moon et al. 2005), probably due to the continuously mode of operation of the biosensor.

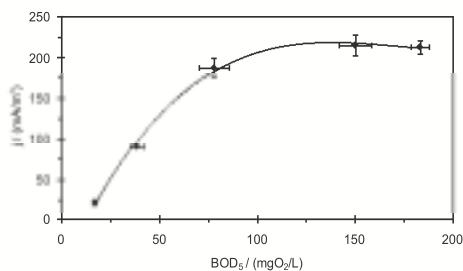


Figure 1: Current density ( $j$ ) generated in the SMFC as a function of initial BOD<sub>5</sub> concentration of wastewater.

#### Effect of environmental conditions

The effect of changes in environmental parameters in the performance of the biosensor was assessed using domestic wastewater with a BOD<sub>5</sub> of  $144 \pm 9 \text{ mg O}_2/\text{L}$  (Figure 2).

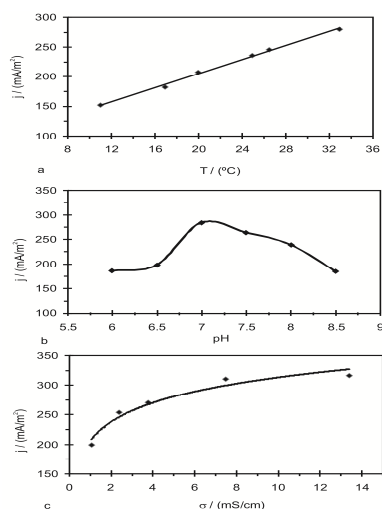


Figure 2: Effect of temperature (T) (a), pH (b) and conductivity ( $\sigma$ ) (c) on current density ( $i$ ) generation.

The  $j$  increased linearly with T (Figure 2a), about  $6 \text{ mA}/(\text{m}^2 \text{ } ^\circ\text{C})$ . The maximum  $j$  ( $288.33 \text{ mA}/\text{m}^2$ ) was obtained at pH 7.0, while the minimum  $j$  correspond to pH interval  $6 \pm 0.1 - 6.5 \pm 0.1$  and to pH  $8 \pm 0.1$  with respectively  $185.55 \text{ mA}/\text{m}^2$  and  $184.44 \text{ mA}/\text{m}^2$  (Figure 2b). These results are consistent with literature studies regardless of MFC configuration and substrates (Gil et al. 2003). The  $j$  increased with the  $\sigma$  of domestic wastewater in the range of  $1.1 \pm 0.012 \text{ mS}/\text{cm}$  to  $7.51 \pm 0.01 \text{ mS}/\text{cm}$ , from  $199 \text{ mA}/\text{m}^2$  to  $316 \text{ mA}/\text{m}^2$ , respectively, the effect being more important between  $1.1 \pm 0.012$  and  $2.1 \pm 0.013 \text{ mS}/\text{cm}$  (Figure 2c). These results suggested that a correction factor should be applied to measurements done under other environmental conditions.

#### CONCLUSIONS

BOD<sub>5</sub> values of up to  $78 \pm 8 \text{ mg O}_2/\text{L}$  could be measured based on a linear relation. pH and Temperature influenced the current densities. The optimum pH was 7. The advantage of the SMFC is that no special anaerobic chamber (anode chamber) is needed because the sensor might be directly submerged in a wastewater channel or anaerobic reactor.

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